

## Experimentation and numerical analysis of biofuel applied in automotive fuel tank

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**Abstract** - Modern trends emission control in this automotive world successfully introduced alternative fuels, as a biofuel is getting more significance to replace diesel fuel entirely or moderately. However, big challenge in corrosion of advanced materials in biodiesel is a major concern as this can reduce vehicle efficiency. This research focused to investigate the corrosion deformation of carbon steeled material at 10 different temperatures such as atmospheric temperature, 50 and 60 °C. Stabling absorption tests in pure diesel, biodiesel in 50% diesel, pure biodiesel were carried out for 720hr.past literature only realized techniques was upgraded. At the completion of the tests, corrosion performance quality's was examined by weight loss measurements techniques exposed metal surface. Biofuel's were analyzed by using computational fluid dynamic oxidation analyzer and Microscopic in mandate to investigate the change in acidity and oxidation of fuel respectively upon exposure in different bio fuels inspected phase one. After this proto type Fuel tank unfolded Surface was examined by optical SEM images dispersive Gaping inspecting. Corrosion products were inspecting by SEM techniques also comparing with simulation contour Results monitoring that the corrosion of automotive mild steel material increase of worm heating, the water content and oxidation products are increased or decreased are evaluating our work will processing simulation with the experiment.

**Index words** - Corrosion, Biofuel, Fuel tank.

### I. INTRODUCTION

Currently, fossil fuels are the main resources of energy meeting the world requirements. The fossil-based resources, such as gasoline, petro-diesel and natural gas are limited and insufficient for the future world's energy demands.

### II. LITERATURESURVEY

1. B. Singha, John Korstadb, Y.C. et al. [5]reported to the currently designed engines use gasoline or petro diesel as fuel. Alternative fuels such as biodiesel raise the issue of corrosion in CI engine parts that come into contact with the fuel. This review demonstrates that corrosion is higher with biodiesel than petro diesel fuel. The rate of corrosion is influenced by temperature, water content, microbial growth, and type of feedstock used for synthesis of biodiesel. Feedstock with higher concentrations of unsaturated fatty acids has greater oxidation rates. The commonly employed metals such as aluminum, copper, copper alloys, and steel to some extent, were found to be prone to corrosion. Pitting corrosion has been common in the metals that have been tested for their corrosiveness in biodiesel. However, stainless steel was found to be immune to pitting corrosion. Corrosion was found to cause swelling in the elastomers. The polar nature of the elastomers dissolved in the biodiesel caused the corrosion

2. H.B..Shao, J.M.Wang et al. [7]said that the corrosion process of aluminum in biodiesel follows a behavior similar to that encountered when aluminum is exposed to an aqueous or ethanol alkaline solution. The corrosion current density and free corrosion potential change as a function of the purity of the biodiesel. The results also show that the polarization curves and impedance data were strongly affected by the ohmic potential drops from the electrolyte. The electrochemical data provide a quantitative indicator of the biodiesel quality, and they can potentially substitute the ASTM D130 standard. Further work is

being performed with the aim of exploring the potential use of aluminum corrosion measurement as a quality determination of biodiesel.

3. D.F. Aktas, J.S. Lee et al. [8] said that the local current distribution and corrosion on carbon steel in biodiesel due to water contamination was monitored and mapped for the first time using the WBE technique. While most carbon steel electrodes corroded when exposed to ion-containing tap water, the cathodes were mainly formed at the biodiesel–water interface. The anodic current distribution in water showed a positive correlation with the distance from the biodiesel–water interface and the biodiesel concentration gradient in water. With the fast growth of biodiesel industry, these findings from the WBE method offer new ways to characterize the metal corrosion in biodiesel and provide insights in developing corrosion prevention strategies. The technique can also be used to study the heterogeneous electrochemistry on materials exposed to multiphase systems.

4. Katherine M. Richard et al. [14] said that the global use of biodiesel fuel blends derived from fatty acid methyl esters (FAME) is increasing; driven by legislation derived from political, economic and environmental factors. The presence of FAME biodiesel changes the operating environment of the engine and after treatment devices, affecting the performance characteristics and requirements of the lubricant. As part of a wider research project into the impact of biologically-sourced fuels on crankcase lubricant performance, this paper documents the impact of biodiesel on corrosion-related performance. The effect of FAME biodiesel on lubricant corrosion control and the differences in performance due to FAME source are described. Mechanistic studies into the corrosive nature of FAME are reported. Novel lubricant technologies tailored to control the negative impact of FAME in the crankcase are demonstrated. It has been observed that the corrosion performance of diesel engine oils is negatively affected by the presence of FAME. This behavior is seen across multiple commercial heavy-duty diesel oils. The natural source of the FAME has an impact on the corrosivity, with highly saturated PME having the least impact. It has been shown that the number of double bonds in the fatty acid chain correlates with the FAME-induced corrosion.

### III. EXPERIMENTAL PROCEDURE

In order to perform experimental tests, the engine was connected to an Eddy Current dynamometer. The objective is to simulate realistic vehicle operating conditions in the test bed. As shown in Fig. 3.1, two axial fans with accompanied ducting were employed to generate the air flow over moving vehicle. The engine was SI type, L-4, 1.3 liter with a multi-point fuel injection (MPFI) system. The Fuel tank was a three-way Fuel tank model in which its properties are summarized in Table 2. The thickness of NiCr thermocouples used in the experiment was 1 mm. Several thermocouples protruded their heads 5 mm out of the channels for measuring the inflow gas temperature. Along radial direction, they were located at the positions with coordinates of 0.2R, 0.5R and 1R. The differential pressure transducer was a digital type with diaphragm sensor calibrated between 0–50 mbar. Exhaust gas velocity was measured by vane. Torque and engine speed were measured by the Eddy Current dynamometer. Pollutant concentrations, namely CO and HC were measured by an AVL model 4000 gas analyzer before and after Fuel tank in order to determine its conversion efficiency in different conditions of engine load and engine speed. A schematic installation of thermocouples, differential pressure transducer, and gas analyzer is shown in fig. The Fuel tank under the test is shown in Fig



Fig 1 Experimental setup

### IV. RESULT AND DISCUSSION

Exhaust Air velocity was measured by vane. Torque and engine speed were measured by the Eddy Current Sensor. Pollutant concentrations, namely Hight were measured by an SII model 2000 Biofuel analyzer before and after Fuel tank in order to determine its conversion efficiency in different conditions of engine load and engine speed.

Solution ID #	Mass change at 1 month (g)	Mass change at 2 months (g)	Mass change at 3 months (g)
STI-1	-0.1014 ± 0.0919	-0.0280 ± 0.0045	-0.0353 ± 0.0077
STI-2	0.0053 ± 0.0026	-0.0059 ± 0.0173	0.0091 ± 0.0007
STI-3	-0.0144 ± 0.0697	0.0131 ± 0.0099	0.0138 ± 0.0117
STI-4	0.0067 ± 0.0458	-0.0480 ± 0.0835	-0.0084 ± 0.0235
STI-5	-0.0227 ± 0.0309	-0.00175 ± 0.0008	-0.0005 ± 0.0008
STI-6	0.0072 ± 0.0044	-0.0126 ± 0.0556	-0.0455 ± 0.0612
STI-7	0.0013 ± 0.0047	0.0063 ± 0.0090	0.0054 ± 0.0094
STI-8	-0.0195 ± 0.0158	0.3100 ± 0.4397	-0.0171 ± 0.0291
STI-9	-0.0075 ± 0.0025	0.0578 ± 0.0873	-0.0004 ± 0.0023
STI-10	-0.0405 ± 0.0479	-0.0429 ± 0.0532	-0.0392 ± 0.0571
STI-11	0.4078 ± 0.5791	0.0060 ± 0.0035	0.0090 ± 0.0008
STI-12	-0.0048 ± 7.07 × 10 <sup>-3</sup>	0.0037 ± 0.0041	0.0057 ± 0.0023
STI-13	-0.0029 ± 0.0031	-0.0354 ± 0.00580	-0.0342 ± 0.0016
STI-14	-0.0019 ± 0.0064	0.0024 ± 0.0002	0.0048 ± 0.0001
STI-15	-0.3216 ± 0.4890	0.0028 ± 0.0044	0.0018 ± 0.0006
STI-16	-0.0066 ± 0.0112	-0.0265 ± 0.0339	-0.0218 ± 0.0342



Fig2 Experimentation Analysis Visual Effects

Table 1 Experimentation Analysis.

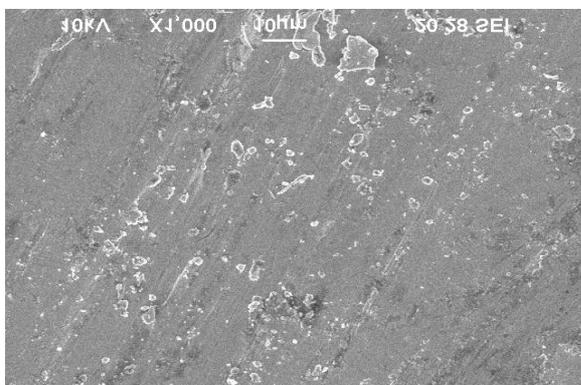
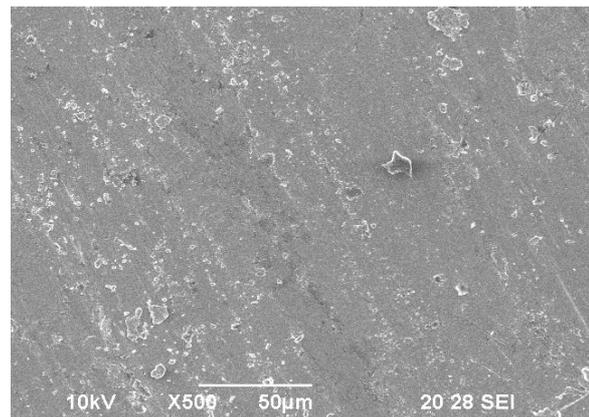
Fig 3 Scanning Electron Microscope (SEM) -Biofuel  
– Neem B50

Fig 4 The SEM images of Mild Steel 50µm × 500

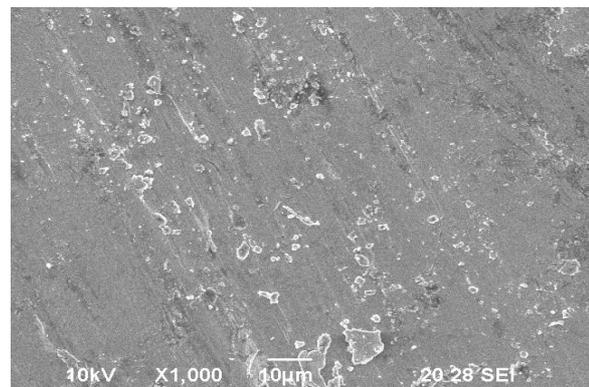


Fig 5 The SEM images of Mild Steel 10µm × 1000

From the Fig 5 it was observed that there will be a fine film is visible which reduce the corrosion of the material that means it increase the lubricating property of the material. The black portions represent the MoS<sub>2</sub>. Moreover the SEM image shows the uniformity in the distribution.

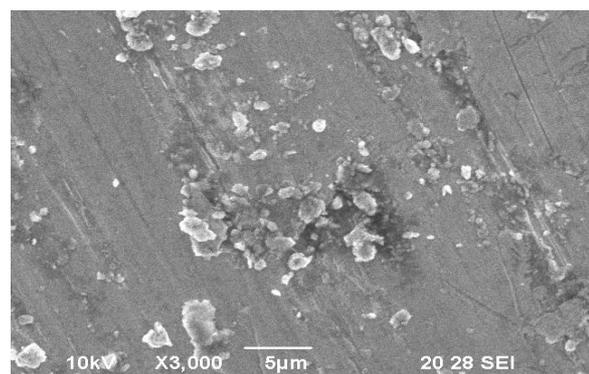


Fig 6 The SEM images of Mild Steel 5µm × 3000

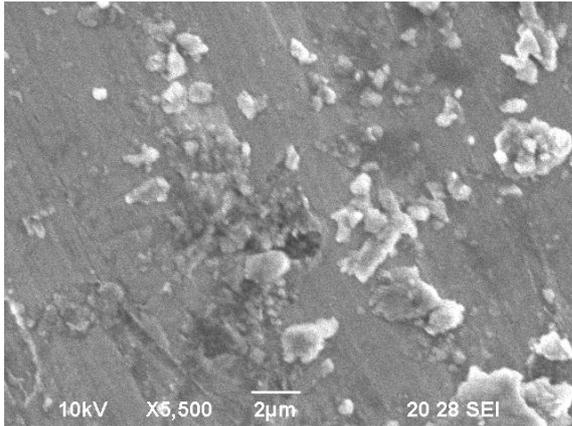


Fig 7 The SEM images of Mild Steel 2µm × 5500

The image shows the unevenness of the specimen and hence the white patches here and there are seen. During SEM the hard particles creates voids which are seen in the image

#### V.CONCLUSION

Reinforcement improves the Mild steel material hardness to an extent of 15% when compared with base Mild steel Material bonding. The Tensile strength improves to an extent of 21% due to the addition of Mild Steel. Results from SEM convey the presence of reinforcement and uniform distribution. Finally comparing numerical results are to consumed old cause material in Aluminum alloy vs without coating material with coating alloy to comparing thermal bonding material should be very quickly and effectively and without failure to be consist to apply bonded material causes in Better performance analysis in Aluminum causes material in room temperature perfectly bonded in different load conditions in natural thermal conduction.

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